

УДК 552 778.3, 665.521.8

## RESEARCH OF RADIATION-CHEMICAL TRANSFORMATION OF SYNTHETIC OIL FROM OIL-BITUMINOUS ROCK

**L.Y.Jabbarova, I.I.Mustafaev, Z.O.Nabizade, R.S.Rzayev, N.A.Ibadov**

*Institute of Radiation Problems of the National Academy of Sciences of Azerbaijan  
AZ 1143, Baku F.Agayev str. 9; e-mail:Clala@mail.ru*

*Synthetic oil from oil-bituminous rocks of the Balakhany deposit has been used as a test subject. The regularities of synthetic oils radiation-chemical transformations have been studied. Laboratory studies have been carried out in an interval of absorbed dose 43-216 kGr and dose rate  $P=0.5$  Gr/s ( $\gamma$  - radiation). Concentration and radiation-chemical yields of obtained gases in various absorbed doses established. Results of studies make it possible to estimate the possible use of oil products obtained from synthetic oil in terms of isolation of radioactive sources from environment.*

**Keywords:** bituminous rock, synthetic oil, radiation-chemical transformations, radiolysis.



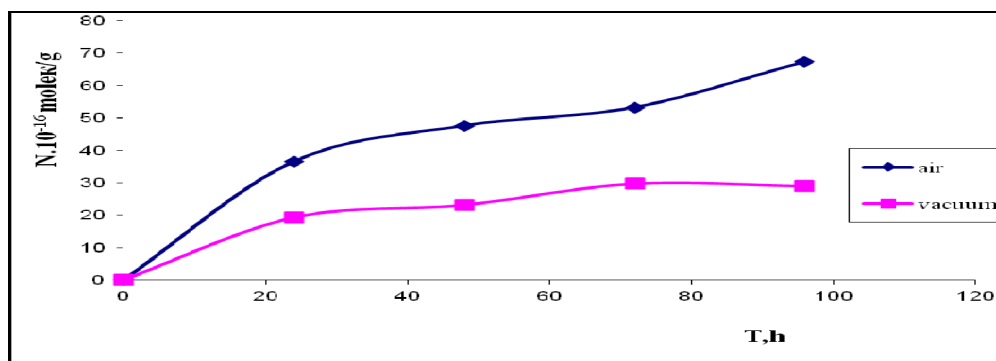
Bituminous rock (BR) - the natural material formed from oil in the top layers of earth crust due to the slow evaporation of easy fractions from it, natural de-asphalting of oil, as well as processes of interaction of its components with oxygen and sulfur. According to the UNO, world geological resources of bituminous rocks (BR) make up approx. 360 billion tons due to the recalculation of a hydro-carbonic component as an alternative source of hydro-carbonic raw materials. Total world oil extraction from BR makes up about 84 million barrels per day. BR resources in Azerbaijan make up 200 million tons.

**The purpose** of this work is to examine the radiation stability of synthetic oil from BR.

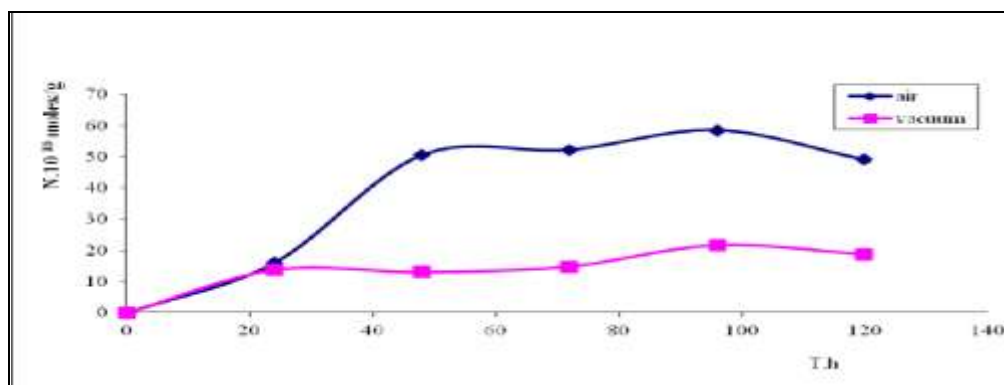
Results of the study will allow to estimate possibility of the production of various oil products from synthetic oil by radiation-chemical methods, as well as the use of these materials in terms of isolation of radioactive sources from environment. We analyzed the synthetic oil obtained from BR on the Balakhany deposit, Azerbaijan. Through distilling on Retort Heating Jacket device at temperature 950 F (510°C) from 375 g we have obtained 50 ml synthetic oil. Rock composition is as follows: oil – 22%, water – 6%, sand – 2%. Experiments have been carried out in  $\gamma$  – source  $^{60}\text{Co}$  at dose rate of 0,5 Gy/s and in absorbed doses of 43-216 kGy. Gas products are analyzed by a gas chromatography method.

## EXPERIMENTAL RESULTS

On Fig. 1 as an example is shown a kinetic curve of formation gases in the vacuum and in the presence of air during gamma-radiolysis of synthetic oil from bituminous rock.



a



b

**Fig. 1 (a,b).** Kinetic curve of methane (a) and ethane (b) accumulation during gamma-radiolysis of synthetic oil from natural bitumens.  $P=0.5$  Gy/s.

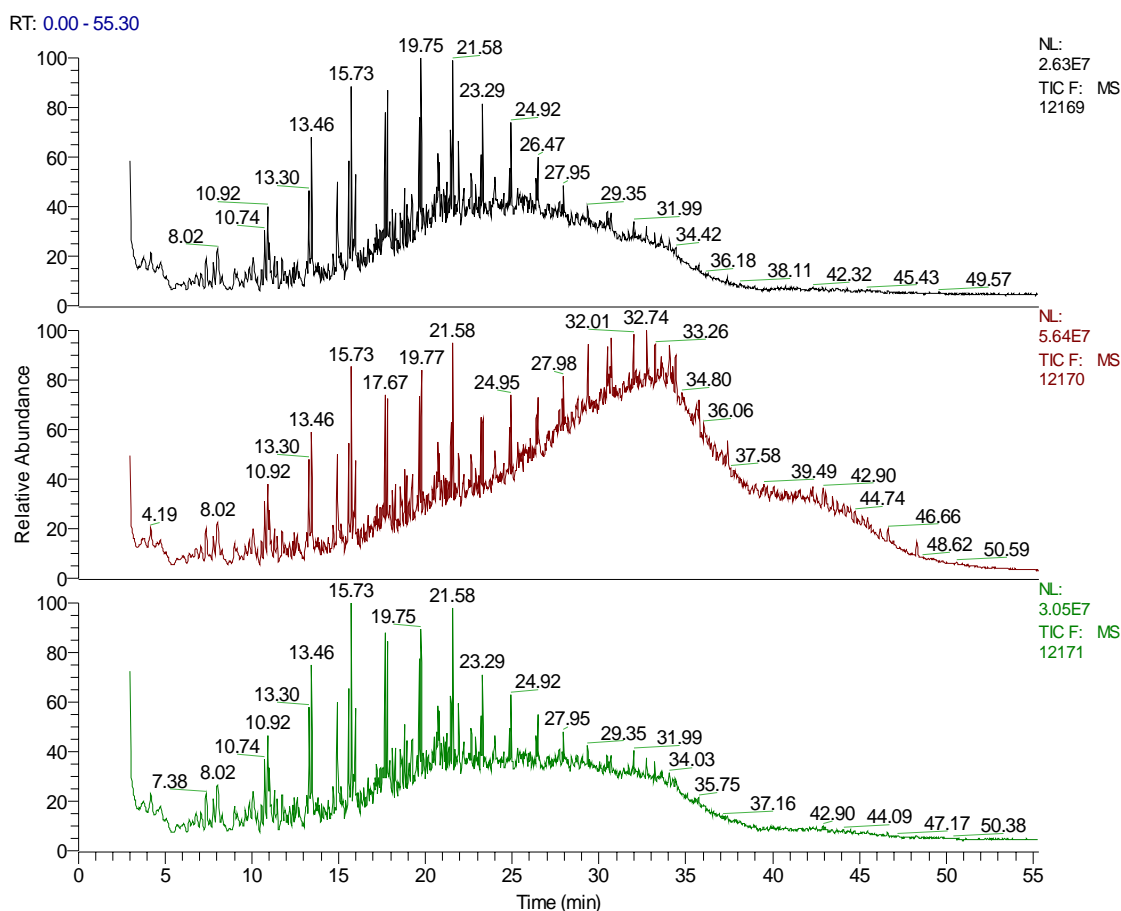
As is known, in all gases the oxygen acts as accelerating factor in the process of radiation-chemical decomposition of synthetic oil which is related to the behavior of

oxidizing-destructive reactions with the participation of radiolytic radicals. Average values of radiation-chemical yields of gases are shown in Table 1.

**Table 1.** Average values of radiation-chemical yields of gases (molec/100 eV) synthetic bituminous oil

Average values of radiation-chemical yields of gases (molec/100 eV) from BR	H <sub>2</sub>	CO	CO <sub>2</sub>	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>2</sub> H <sub>4</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>	C <sub>7</sub>
Air	0.31	0.85	0.66	0.06	0.05	0.024	0.13	0.33	0.32	0.16	0.04
Vacuum	0.37	0.26	0.67	0.03	0.02	0.005	0.014	0.022	0.176	0.13	0.06

Fig. 2 provides chromatograms samples of components in the input and irradiated samples synthetic oil. According to chromatograms, of synthetic oil are identified



**Fig. 2.** Chromatograms0 in the input and irradiated synthetic oil in air and vacuum

Identified components of input synthetic oil are shown in Table 2.

**Table 2.** Identified components of input synthetic oil

<i>N</i> <sub>o</sub>	RT, min	Identified components of input synthetic oil	Formula
1	4,17	Toluene	C <sub>7</sub> H <sub>8</sub>
2	7,37	p-Xylene	C <sub>8</sub> H <sub>10</sub>
3	7,79	cis-2-Nonene	C <sub>9</sub> H <sub>18</sub>
4	8,02	Nonane	C <sub>9</sub> H <sub>20</sub>
5	9	Octane, 2,6-dimethyl-	C <sub>10</sub> H <sub>22</sub>
6	9,65	1-Octyn-3-ol, 4-ethyl-	C <sub>10</sub> H <sub>18</sub> O
7	10,09	Benzene, 1-ethyl-3-methyl-	C <sub>9</sub> H <sub>12</sub>
8	10,56	Benzene, (1-methylethyl)-	C <sub>9</sub> H <sub>12</sub>
9	10,74	1-Decene	C <sub>10</sub> H <sub>20</sub>
10	10,92	Decane	C <sub>10</sub> H <sub>22</sub>
11	11,31	1-Octanol, 2-methyl-	C <sub>9</sub> H <sub>20</sub> O
12	11,75	Benzene, 1-ethyl-4-methyl-	C <sub>9</sub> H <sub>12</sub>
13	13,3	1-Undecanol	C <sub>11</sub> H <sub>24</sub> O
14	13,46	Undecane	C <sub>11</sub> H <sub>24</sub>

15	14,67	Undecane, 6-methyl-	C <sub>12</sub> H <sub>26</sub>
16	14,93	Benzene, 1,2,4,5-tetramethyl-	C <sub>10</sub> H <sub>14</sub>
17	15,58	Cyclopropane, nonyl-	C <sub>12</sub> H <sub>24</sub>
18	15,73	Dodecane	C <sub>12</sub> H <sub>26</sub>
19	15,97	Undecane, 2,6-dimethyl-	C <sub>13</sub> H <sub>28</sub>
20	17,67	1-Tridecene	C <sub>13</sub> H <sub>26</sub>
21	17,8	Tridecane	C <sub>13</sub> H <sub>28</sub>
22	18,81	1-Pentadecanol	C <sub>15</sub> H <sub>32</sub> O
23	18,92	Cyclohexanol, 5-methyl-2-(1-methylethyl), [1R-(1à,2á,5à)]-	C <sub>10</sub> H <sub>20</sub> O
24	19,64	1-Tetradecene	C <sub>14</sub> H <sub>28</sub>
25	19,75	Tetradecane	C <sub>14</sub> H <sub>30</sub>
26	20,73	Naphthalene, 2,7-dimethyl-	C <sub>12</sub> H <sub>12</sub>
27	20,81	Tetradecane, 2,6,10-trimethyl-	C <sub>17</sub> H <sub>36</sub>
28	21,45	2,6-Dodecadien-1-ol, 3,7,11-trimethyl,(E,E)-	C <sub>15</sub> H <sub>28</sub> O
29	21,58	Pentadecane	C <sub>15</sub> H <sub>32</sub>
30	21,92	1H-Indene, 2,3,3a,4,7,7a-hexahydro-2,2,4,4,7,7-hexamethyl	C <sub>15</sub> H <sub>26</sub>
31	22,64	Naphthalene, 1,6,7-trimethyl-	C <sub>13</sub> H <sub>14</sub>
32	23,19	1-Hexadecanol	C <sub>16</sub> H <sub>34</sub> O
33	23,29	Hexadecane	C <sub>16</sub> H <sub>34</sub>
34	24,84	1-Heptadecanol	C <sub>17</sub> H <sub>36</sub> O
35	24,92	Heptadecane	C <sub>17</sub> H <sub>36</sub>
36	26,4	8-Heptadecene	C <sub>17</sub> H <sub>34</sub>
37	26,47	Octadecane	C <sub>18</sub> H <sub>38</sub>
38	27,95	Nonadecane	C <sub>19</sub> H <sub>40</sub>
39	29,37	Eicosane	C <sub>20</sub> H <sub>42</sub>
40	30,49	Octadecane, 3-methyl-	C <sub>19</sub> H <sub>40</sub>
41	30,72	Heneicosane	C <sub>21</sub> H <sub>44</sub>
42	32,74	Allopregnane	C <sub>21</sub> H <sub>36</sub>
43	33,26	Octadecane, 3-ethyl-5-(2-ethylbutyl)-	C <sub>26</sub> H <sub>54</sub>

To analyze the composition of source <sup>60</sup>Co at a dose rate of 0,5 Gy/s and an bituminous oil we divided its distillation into 3 absorbed dose of 86.4 kGy. Radiation-fractions: up to 110°C; 125°C and 145°C. chemical yields of gases are summarized in Samples of fractions have irradiated on a γ – Table 3.

**Table 3.** Radiation-chemical yields of gases (molec/100 eV) in bituminous synthetic oil at absorbed dose of 86.4 kGy, P=0.5 Gy/s

Fractions from BR, °C	H <sub>2</sub>	CO	CO <sub>2</sub>	CH <sub>4</sub>	C <sub>2</sub> H <sub>6</sub>	C <sub>2</sub> H <sub>4</sub>	C <sub>3</sub>	C <sub>4</sub>	C <sub>5</sub>	C <sub>6</sub>	C <sub>7</sub>
1. T< 110	0.70	1.96	0.38	0.91	0.22	0.02	0.04	0.01	0.05	0.28	0.23
2. 110<T<125	0.42	3.72	0.89	0.60	0.41	0.03	0.07	0.04	0.28	0.38	0.14
3. 125<T<145	0.67	3.51	2.18	0.24	0.53	0.08	0.09	0.02	0.09	0.13	0.05

As is seen from Table 3, the total radiation-chemical yields of gases from fraction 1 are as follows: - 4.8 molec/100 eV, from fraction 2 - 6.9 molec/100 eV and from fraction 3 - 7.5 molec/100 eV which is due to the concentration and differences in molecular structure of the compounds making up a part of these fractions. It is necessary to note that paraffin and polycyclic aromatic hydrocarbons have relatively high stability against radiation influence. At the same time, functional groups, especially oxygen-containing, as well as olefinic hydrocarbons, less radio-immunity. Stability of these organic compounds against radiation depends on the potential of excited

conditions and ionization, i.e. all what defines behavior of energy transmission processes between components. In the presence of the poly-conjugate aromatic structures, the absorbed energy dissipates  $\pi$ -electrons, and as a result, bond breakage occur in functional groups. The irradiation of these samples in the air is brought to a small growth of destruction process, however, yields of products remain relatively low. To increase the radiation-chemical yield of gases and activate the chain mechanism of hydrocarbons' decomposition in these systems, it is necessary to apply high temperatures.

### CONCLUSION

Relatively high radiation stability of synthetic oil from BR in the vacuum and the air is attributable to the presence of paraffin and polycyclic aromatic hydrocarbons in its structure. It has been established that the products obtained from BR are noted for high stability against the influence of radiation and temperature of 50°C. Synthetic oil can be used as feed stock for the production of waterproof

materials to be applied in terms of radiation influence. Organic part of BR can serve as a perspective source for obtaining of various kinds of fuels, oils, coke, and bitumen. To obtain the hydrogen, hydrocarbon gases and olefinic hydrocarbons from synthetic oil-combined influence of ionizing radiation and temperature are required at the agreed value of temperature and capacity of radiation.

### REFERENCES

1. Джаббарова Л.Ю. Радиационно-термические превращения битуминозных пород. Дисс.. канд.хим.наук. Институт Радиационных Проблем НАНА. 2007. 152 с.
2. Mustafaev I.I., Jabbarova L.Y., Yagubov K.M., Gulieva N.G. Journal of Radiation Safety Problems in the Caspian Region, Kluwer Academic Publishers. Printed in the Netherlands. 2004. p. 141-146.
3. Mustafaev I.I., Jabbarova L.Y., Gulieva N.G., Yagubov K.M. //Journal of Radioanalytical and Nuclear Chemistry. Budapest, Akademia Kiado, vol. 262. 2004. №2. p. 509-511.

### **ИССЛЕДОВАНИЕ РАДИАЦИОННО-ХИМИЧЕСКОГО ПРЕВРАЩЕНИЯ СИНТЕТИЧЕСКОЙ НЕФТИ ИЗ НЕФТЕБИТУМИНОЗНОЙ ПОРОДЫ**

**Л.Ю.Джаббарова, И.И.Мустафеев, З.О.Набизаде, Р.С.Рзаев, Н.А.Ибадов**

*В качестве объекта исследования использовалась синтетическая нефть из битуминозных пород Балаханского месторождения. Изучены закономерности радиационно-химического превращения синтетической нефти. Лабораторные исследования проводились в интервалах поглощенной дозы 43-216 кГр и мощности дозы  $P=0.5$  Гр/с ( $\gamma$ -излучения). Установлены концентрация, радиационно-химические выходы полученных газов при различных поглощенных дозах. Результаты*

исследований позволяют оценить возможность использования нефтепродуктов, полученных из синтетической нефти, для изоляции радиоактивных источников из окружающей среды.

**Ключевые слова:** нефтебитуминозная порода, синтетическая нефть, радиационно-химические превращения, радиоллиз.

### **NEFT-BİTURLU SÜXURLARDAN ALINMIŞ SİNTETİK NEFTİN RADİASIYA-KİMYƏVİ ÇEVRİLMƏLƏRİNİN TƏDQIQI**

**L.Y.Cabbarova, İ.İ. Mustafayev, Z.O.Nəbizadə, R.S.Rzayev, N.A.İbadov**

*Tədqiqat obyektı olaraq alternativ enerji mənbəyi olan Balaxanı neft-bitum süxurlarından alınmış sintetik neftdən istifadə olunmuşdur. Sintetik neftin radiasiya-kimyəvi çevrilmələrinin qanunauyğunluqları öyrənilmişdir. Laboratoriya tədqiqatları udulan dozanın 43-216 kQr intervalında və  $P=0,5$  Qr/s doza gücündə ( $\gamma$ -şüalanma) aparılmışdır. Müxtəlif udulma dozalarında alınmış qazların qatılığı, radiasiya-kimyəvi çıxımı təyin edilmişdir. Tədqiqatların nəticələri sintetik neftdən alınmış neft məhsullarının radioaktiv mənbələrin ətraf mühitdən izolyasiya olunmasında istifadə olunma mümkünlüyünü qiymətləndirməyə imkan verir.*

**Açar sözlər:** neft-bitum süxuru, sintetik neft, radiasiya-kimyəvi çevrilmə, radioliz.

*Redaksiyaya daxil olub 30.01.2013.*