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SYNTHESIS AND CRYSTAL STRUCTURE OF A NEW 9P-TYPE LAYERED van der WAALSCOMPOUND SnBi_4Te_4 ¹E.N. Orujlu, ¹A.E. Seidzade, ^{2,3}Z.S. Aliev, ²I.R. Amiraslanov, ¹M.B. Babanly¹Acad. M. Nagiyev Institute of Catalysis and Inorganic Chemistry of ANAS, 113, H. Javid ave., AZ 1143, Baku, Azerbaijan, e-mail: elnur.oruclu@yahoo.com²G.M. Abdullayev Institute of Physics of ANAS, 131, H. Javid ave., AZ 1143, Baku, Azerbaijan³Azerbaijan State Oil and Industry University, 20, Azadlig ave., AZ 1010, Baku, Azerbaijan

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Abstract: Considering structural features of the already known tetradymite-like layered chalcogenide phases, we have attempt to synthesize a new mix-layered compound, SnBi_4Te_4 . The newly synthesized alloy examined by means of differential thermal analysis, powder X-ray diffraction and scanning electron microscope techniques. The melting nature of the discovered phase is found to be incongruently at 831 K. The crystal structure of the SnBi_4Te_4 was elucidated from powder pattern by Rietveld method. The determined crystal structure was found to derived from tetradymite type and featured by the alternation of theseven-layered (septuple) blocks of SnBi_2Te_4 and bismuth bilayers. The result of this work, the existence of a new - SnBi_4Te_4 compound in the Sn–Bi–Te ternary system can shed light for the incoming research works to search for similar phases in the other related $A^{\text{IV}}\text{–Bi–Te}$ ($A^{\text{IV}} = \text{Ge, Sn, Pb}$) systems.

Keywords: Sn–Bi–Te ternary system, SnBi_4Te_4 , novel layered chalcogenide, Bi-bilayers, van der Waals compounds, crystal structure, Rietveld method, topological insulator.

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Introduction

The layered van der Waals (vdW) chalcogenides, in particular, tellurides of the group 15 metals have been extensively studied during the last years as prospective thermoelectric materials and topological insulator for energy conversion and spintronic applications [1-3]. Tuning the band electronic properties of these materials are of interesting from the point of view of their applications in real optoelectronic and spintronic devices. The rational tuning of the electronic properties is possible by the doping, nanostructuring or heterostructuring by combination of various atomic blocks or layers. Thanks to the existing vdW gap in the mentioned materials, inserting the, e.g., atomic bilayers in rocksalt-type septuple blocks are always simple way to modify crystal structure and electronic properties [3-5].

Recently preparation of mixed-layered compounds having thermoelectric properties are considered to be a more effective method

for material design [6-7]. The literature data shows that, the systems $A^{\text{IV}}B^{\text{VI}}\text{–}A_2^{\text{V}}B_3^{\text{VI}}$ ($A^{\text{IV}} = \text{Ge, Sn, Pb}$; $A^{\text{V}} = \text{Sb, Bi}$; $B^{\text{VI}} = \text{Se, Te}$) host tetradymite-type layered ternary compounds and exhibit promising thermoelectric properties [7-14]. On the other hand, the discovery of the topological insulating properties in these compounds made them much more attractive in the last few years [15-20].

The Sn–Bi–Te ternary system [21-23] has been studied by various groups of authors so far [21-23]. According to Karpinski [21], this system includes three stable ternary compounds, namely SnBi_2Te_4 , SnBi_4Te_7 , $\text{SnBi}_6\text{Te}_{10}$ which are melt by peritectic reactions at 873, 863 and 855 K, respectively. Later, Kuropatwa [22] and Chiu [23] independently have reported an additional two ternary compounds - $\text{Sn}_2\text{Bi}_2\text{Te}_5$ and SnBiTe_2 . Literature data show that, in order to overcome metastable state, annealing at higher than 700

K is important in the compositional range of 30 and 53 mol % SnTe [21]. There by, it seems that very hard to obtain further stable homogeneous phases in the SnTe–Bi₂Te₃ system. May be very long annealing time could be helpful, but present reports on this system confirm that the phase diagram of this system is still contradictory and further thoroughly experimental investigations are necessary.

SnBi₂Te₄ is one of the main dominant phases in the SnTe–Bi₂Te₃ system and, its septuple structure is formed by the insertion of SnTe into rocksalt-type slabs of Bi₂Te₃. The crystal structure of this compound is a long-periodically stacking sequence of these septuples along the *c* axes. The crystal structure of the other two compounds SnBi₄Te₇ and SnBi₆Te₁₀ built-up alternation of quintuples of the Bi₂Te₃ and septuples of the SnBi₂Te₄ thus can be considered mixed-layer blocks according to -7-5-7-5-7-5- and -7-5-5-

7-5-5-7-5-5- sequence, respectively. Crystal structure information of these compounds can be found in the Refs[21,24].

The combination of the elemental bismuth or antimony bilayers and quintuple blocks of Bi₂Te₃ (or Sb₂Te₃ and Bi₂Se₃) and the septuple block of the GeBi₂Te₄, PbBi₂Te₄ ternaries are also found to be stable structures [6]. Taking into account alternation of quintuple, septuple and, mixed-layered structures exist in the (A^{IV}Te)_{*n*}·(Bi₂Te₃)_{*m*} (A^{IV} = Ge, Sn, Pb) homologous series, there is a possibility to design (A^{IV}Te)_{*n*}·(Bi₂Te₃)_{*m*}·(Bi₂)_{*k*} layered phases where there is an alternation of *k*Bi bilayers, *n*A^{IV}Te quintuples and *m*A^{IV}Bi₂Te₄ septuples.

Here we present the synthesis and elucidation of a crystal structure of 9*P*-type heterostructured new ternary compound - SnBi₄Te₄, which is consists of alternating bismuth bilayers and SnBi₂Te₄ septuple packets. The melting nature and temperature were also presented here.

Experimental part

Elemental Tin, Bismuth, and Tellurium (99.999% purity, Alfa Aesar company) were used as starting components to synthesize polycrystalline SnBi₄Te₄. The stoichiometric mixture of elements was sealed in evacuated (10⁻² Pa) quartz ampule and heated up to 1000 K and kept at this temperature for 5 h and then water quenched. In order to achieve complete homogenization, the sample annealed for ~700 h at 700 K.

The sorted-out ingot was examined by differential thermal analysis and powder X-ray diffraction techniques. DTA measurement was

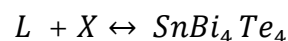
performed using a NETZSCH 404 F1 Pegasus system at a heating rate of 10 K·min⁻¹, while PXRD was done in a Bruker D2 PHASER diffractometer with CuK_α radiation within 2θ = 5-100 range at room temperature. The crystal structure refinement was performed using the EVA and Topas V4.2 softwares by Bruker. The microstructures and equilibrium composition of the title sample was determined by Tescan Vega 3 SBH scanning electron microscope equipped with ThermoScientific UltraDry Compact EDS detector.

Results and discussion

Fig.1 shows the XRD pattern of SnBi₄Te₄ in comparison with initial SnBi₄Te₄ and Bismuth. It is clearly seen that SnBi₄Te₄ has a quite identical diffraction pattern with typical peaks those do not come from initial constituents. The obtained pattern fully indexed with a rhombohedral *P*-3*m*1 (#156) lattice. The crystal structure of the obtained phase was refined by the Rietveld

method and results will be discussed below.

The DTA heating thermogram for SnBi₄Te₄ is given in Fig. 2 with two endothermic effects at 831 K and 906 K. We assume that the first sharp endothermic thermal event corresponding to the peritectic decomposition of SnBi₄Te₄ phase according to reaction.



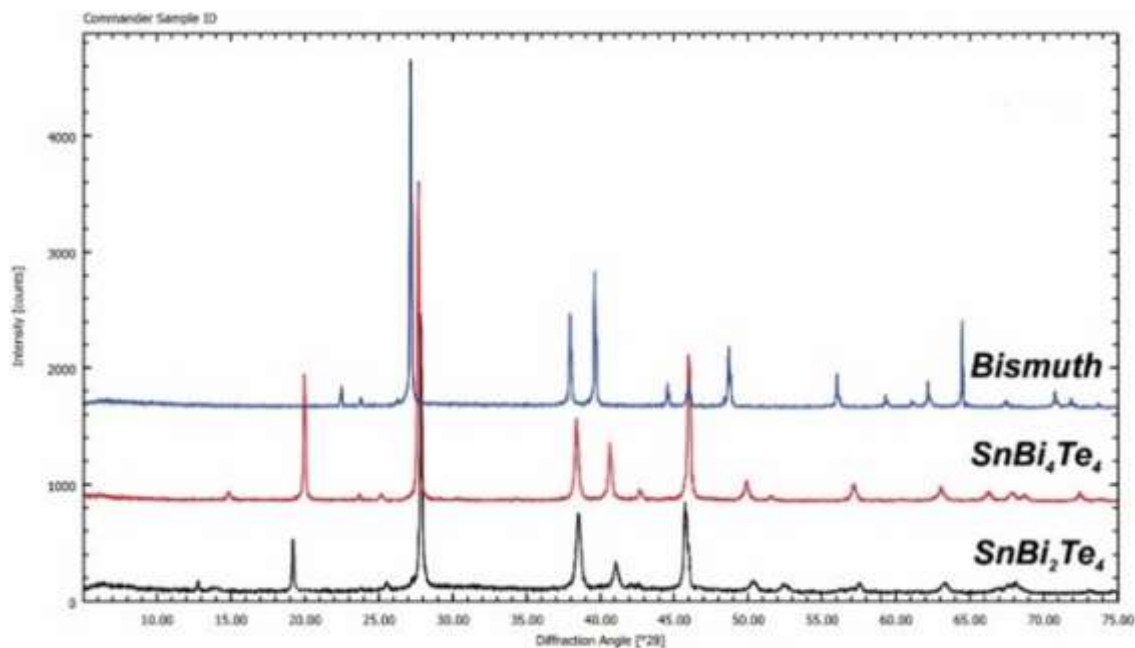


Fig. 1. PXR D patterns of Bismuth, SnBi_4Te_4 , and SnBi_2Te_4 .

The literature data on Sn–Bi–Te ternary system showed that X phase may tin telluride which is in equilibrium with liquid phase at above peritectic temperature. Nevertheless, further experimental results are strongly necessary for the conclusion. The second wide

thermal effect confirms that the melting process starts at the end of the first effect whereas, the effect at 906 K conforms to finish of the melting process and can be considered as liquidus point.

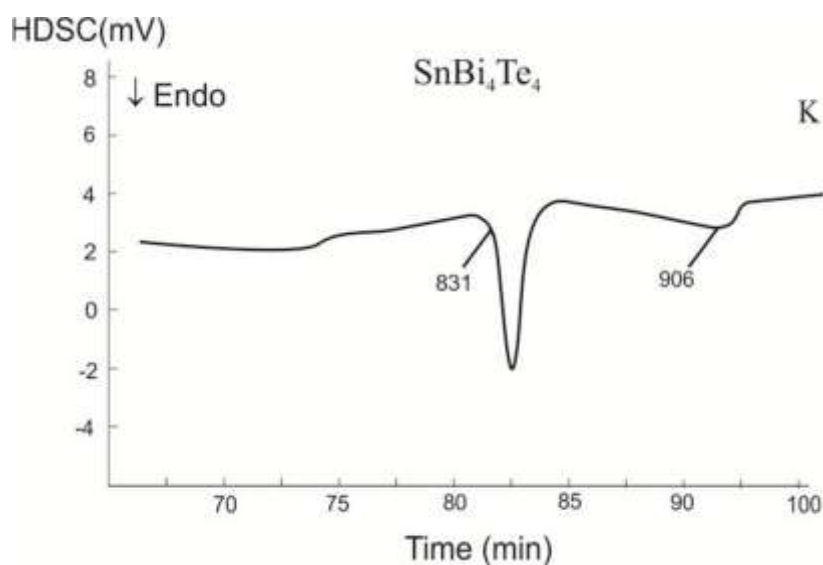


Fig. 2. DTA heating thermogram for the SnBi_4Te_4 .

The further confirmation of the existence and chemical composition of the synthesized alloy comes from SEM-EDS measurements. Fig. 3 illustrates the

homogenous microstructure of the SnBi_4Te_4 alloy, where as its EDS spectrum and equilibrium composition are given in Fig. 4. The layered texture of the alloy is also clearly

seen from micrograph. Obviously, the sample agrees well with the formula SnBi_4Te_4 . chemical composition of the synthesized

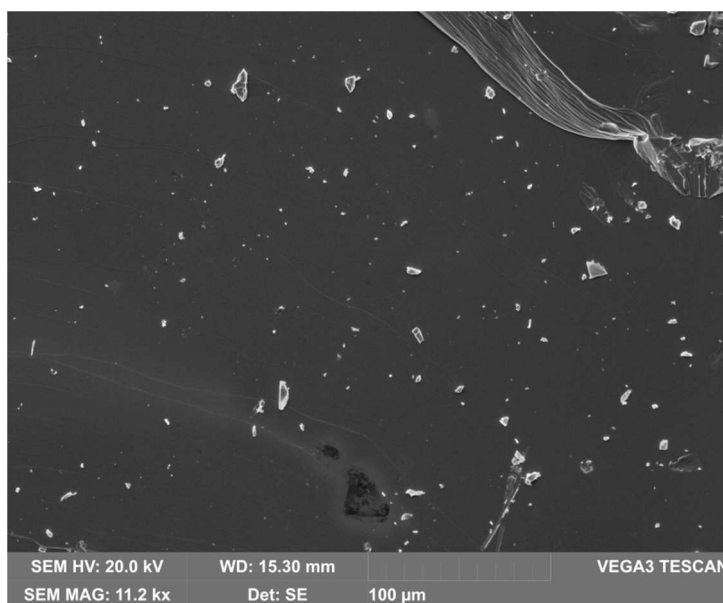


Fig. 3. SEM micrograph of the SnBi_4Te_4 .

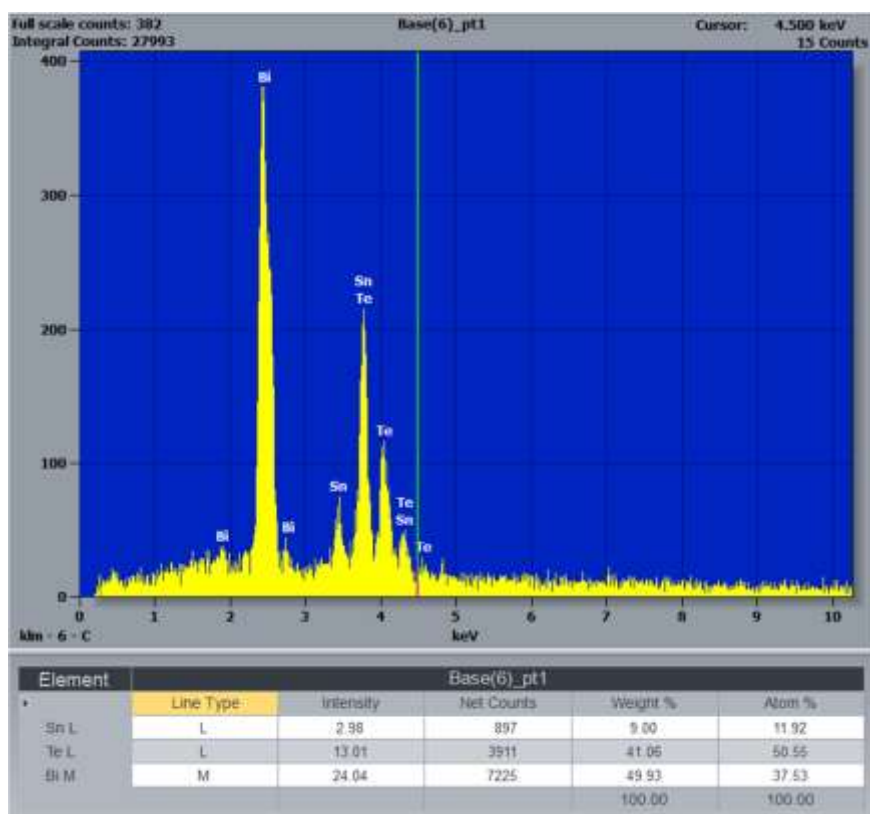


Fig. 4. EDS spectrum and element analysis result for the SnBi_4Te_4 .

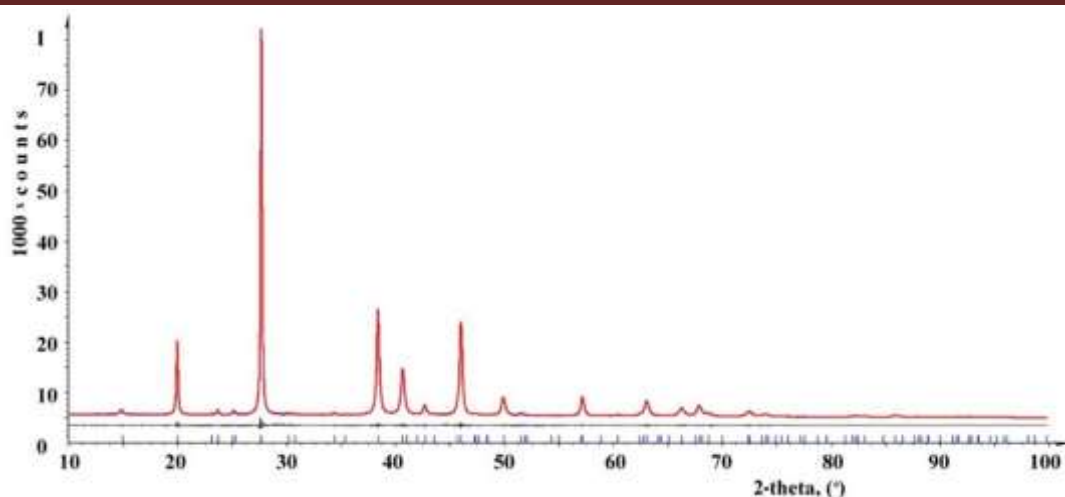


Fig. 5. XRD pattern for the SnBi_4Te_4 .

The crystal structure of the title compound was refined from powder XRD pattern recorded in the range $2\theta = 5$ -100 (Fig. 5). Below the XRD pattern black curve shows

the difference of intensities between the experimental and calculated by Rietveld. A 3D side view of the obtained crystal structure illustrated in Fig. 6.

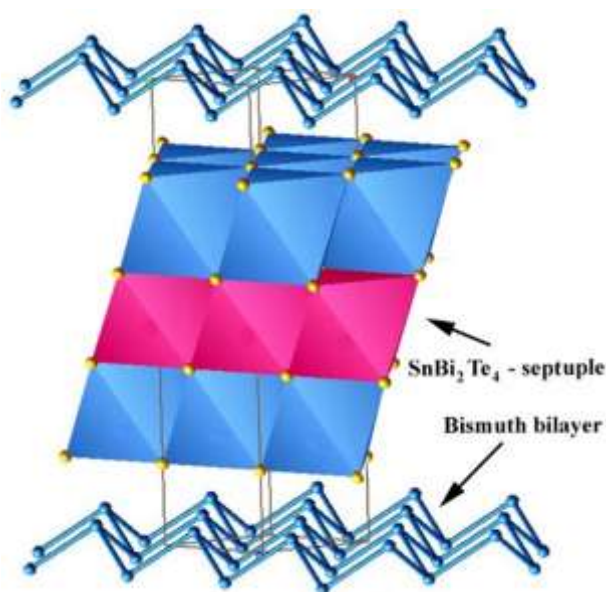


Fig. 6. A 3D side view of the refined crystal structure for the SnBi_4Te_4 .

The refined unit cell parameters, atomic positions, and interatomic distances are shown in Tables 1, 2, and 3.

Table 1. Refined structure parameters for SnBi_4Te_4 .

Space group	<i>P-3m1</i>
Unit cell parameters at 298 K:	
a (Å)	4.43306 (57)
c (Å)	17.7396 (57)
Cell Volume (Å ³)	302.01 (12)
Crystal Density (g/cm ³)	8.0550 (33)
R-Bragg (%)	0.20

Table 2. Atomic positional parameters in SnBi₄Te₄.

Site	N _p	x	y	z	Atom	Occ		B _{eq}
						Single positions	Mixed positions	
Bi1	2	1/3	2/3	0.0451(12)	Bi ⁺³	1	1	1.32
Te1	2	0	0	0.1790(23)	Te	1	1	1.32
Bi2	2	2/3	1/3	0.2706 (13)	Bi ⁺³ Sn ⁺²	1 -	0.76 0.24	1.32
Te2	2	1/3	2/3	0.3938(30)	Te	1	1	1.32
Sn	1	0	0	0.5	Sn ⁺² Bi ⁺³	1 -	0.52 0.48	1.32

Table 3. Interatomic distances in Bi₂·SnBi₂Te₄.

Atoms		Distance
Sn	Te(2)	6 x 3.178(31) Å
Bi(1)	Bi(1)	3 x 3.019(24) Å
Bi(2)	Te(1)	3 x 3.033(25) Å
	Te(2)	3 x 3.366(32) Å

The values presented in Table 2 (except the “mixed positions” column) were obtained for the condition where each crystallographic position was completely occupied by one type of atoms. However, a structural study of compounds containing similar septuple slabs shows that they are characterized by the substitution of metal atoms in atomic layers[25]. Therefore, we also

refined the occupation coefficients in this structure in case of the initial composition of SnBi₄Te₄ is preserved. The occupation results obtained in this case are shown in Table 2, in the “mixed position” column. According to the results of the refinement, the Sn/Bi occupation ratio is found to be 52/48 in the central layer. At the same time, in the Bi(2) site, the Bi/Sn ratio is 76/24.

Conclusion

In this report, we have synthesized a new ternary tetradymite type layered compound - SnBi₄Te₄ in the Sn–Bi–Te ternary system. The obtained phase has a layered structure with rocksalt-type septuple blocks of SnBi₂Te₄ and Bi-bilayers. Thermal analysis

result reveals that the newly found compound melts peritectically at 831 K. The existence of nonuple packets in the Sn–Bi–Te ternary system gives possibility to expect similar phases in the other related A^{IV}–B^V–Te (A^{IV}=Ge, Sn, Pb; B^V=Sb, Bi) ternary systems.

References

- Ivanova L.D., Petrova L.I., Granatkina Yu.V., Zemskov V.S., Sokolov O.B., Ya. Skipidarov S., Kurganov V.A., Podbel'skii V.V. Materials based on bismuth and antimony chalcogenides for thermoelectric cooler stages. *Inorganic Materials*, 2011, vol. 47, no. 5, pp. 459-464.
- Ivanova L.D., Petrova L.I., Granatkina Yu.V., Zemskov V.S., Sokolov O.B., Skipidarov S.Ya., Duvankov N.I. Extruded materials for thermoelectric coolers. *Inorganic Materials*, 2008, vol. 44, no. 7, pp. 687-691.
- Duong D.L., Yun S.J., Lee Y.H. van der Waals Layered Materials: Opportunities and Challenges. *ACS Nano*, 2017, vol. 11,

- issue 12, pp.11803-11830.
- Cecchi S., Dragoni D., Kriegner D., Tisbi E., Zallo E., Arciprete F., et al. Interplay between Structural and Thermoelectric Properties in Epitaxial $\text{Sb}_{2+x}\text{Te}_3$ Alloys. *Advanced functional materials*, 2019, vol. 29, no 2, pp. 1805184.
 - Jiangfeng Ni, Xuanxuan Bi, Yu Jiang, Liang Li, Jun Lu. Bismuth Chalcogenide Compounds Bi_2X_3 (X = O, S, Se): Applications in Electrochemical Energy Storage. *Nano Energy*, 2017, vol. 34, pp. 356-366.
 - Babanly M.B., Chulkov E.V., Aliev Z.S., Shevelkov A.V., Amiraslanov I.R. Phase Diagrams in Materials Science of Topological Insulators Based on Metal Chalcogenides. *Russian Journal of Inorganic Chemistry*, 2017, vol. 62, no. 13, pp. 1703-1729.
 - Shelimova L.E., Karpinskii O.G., Konstantinov P.P., Avilov E.S., Kretova M.A., Zemskov V.A. Crystal Structures and Thermoelectric Properties of Layered Compounds in the $\text{ATe-Bi}_2\text{Te}_3$ (A=Ge, Sn, Pb) Systems. *Inorganic Materials*. 2004, vol. 40, no. 5, pp. 530-540.
 - Kosuga A., Ishibashi H., Kubota Y., Kifune K. Crystal Structure and Thermoelectric Properties of Ge-Sb-Te Homologous Structure. *J. Japan Inst. Met. Mater.* 2015, vol. 79, no. 11, pp. 562-568.
 - Fabian von Rohr, Schilling A., Robert J. Cava Single-crystal growth and thermoelectric properties of $\text{Ge}(\text{Bi}, \text{Sb})_4\text{Te}_7$. *J. Phys. Condens. Matter.*, 2013, vol. 25, no.7, pp. 075804.
 - Li J., Zhang X., Lin S., Chen Z., Pei Y. Realizing the High Thermoelectric Performance of GeTe by Sb-Doping and Se-Alloying. *Chem. Mater.* 2017, vol. 29, pp. 605-611.
 - Kim J.N., Kaviany M., Shim J-H. Optimized ZT of Bi_2Te_3 -GeTe compounds from first principles guided by homogeneous data. *Physical Review B*, 2016, vol. 93, pp. 075119(R).
 - Rosenthal T., Schneider M.N., Stiewe C., Döblinger M., Oeckler O. Real Structure and Thermoelectric Properties of GeTe-Rich Germanium Antimony Tellurides. *Chem. Mater.*, 2011, vol. 23, issue 19, pp. 4349-4356.
 - Wu D., Xie L., Xu X., He J. High Thermoelectric Performance Achieved in $\text{GeTe-Bi}_2\text{Te}_3$ Pseudo-Binary via Van der Waals Gap-Induced Hierarchical Ferroelectric Domain Structure. *Adv. Funct. Mater.*, 2019, pp. 1806613.
 - Omoto T., Kanaya H., Ishibashi H., Kubota Y., Kifune K., Kosuga A. Formation Phases and Electrical Properties of Ge-Bi-Te compounds with Homologous Structures. *Journal of electronic materials.*, 2016, vol. 45, issue 3, pp. 1478-1483.
 - Eremeev S.V., Menshchikova T.V., Silkin I.V., Vergniory M.G., Echenique P. M., Chulkov E.V. Sublattice effect on topological surface states in complex $(\text{SnTe})_{n>1}(\text{Bi}_2\text{Te}_3)_m=1$ compounds. *Phys. Rev. B*, 2015, vol. 91, pp. 245145.
 - Kuroda K., Miyahara H., Ye M., V. Eremeev S., M. Koroteev Yu., Krasovskii E. E., Chulkov E. V., Hiramoto S., Moriyoshi C., Kuroiwa Y., Miyamoto K., Okuda T., Arita M., Shimada K., Namatame H., Taniguchi M., Ueda Y., Kimura A. Experimental Verification of PbBi_4Te_7 as a 3D Topological Insulator. *Phys. Rev. Lett.*, 2012, vol. 108, pp. 206803.
 - Papagno M., Eremeev S.V., Fujii J., Aliev Z.S., Babanly M.B., Mahatha S.K., Vobornik I., Mamedov N.T., Pacilé D., Chulkov E.V. Multiple Coexisting Dirac Surface States in Three-Dimensional Topological Insulator $\text{PbBi}_6\text{Te}_{10}$. *ACS Nano*, 2016, vol. 10, issue 3, pp. 3518-3524.
 - Wang L.L., Johnson D. D. Ternary Tetradymite Compounds as Topological Insulators. *Physical review. B, Condensed matter*, 2011, vol. 83, issue 24, p. 241309.
 - Niesner D., Otto S., Hermann V., Fauster Th., Menshchikova T.V., Eremeev S.V., Aliev Z.S., Amiraslanov I.R., Babanly M.B., Echenique P.M., Chulkov E.V. Bulk and surface electron dynamics in a p-type topological insulator SnSb_2Te_4 . *Physical Review B*, 2014, vol. 89, pp. 081404.

20. Nechaev I.A., Aguilera I., V. De Renzi, A. di Bona, A. Lodi Rizzini, A. M. Mio, G. Nicotra, A. Politano, S. Scalese, Aliev Z.S., Babanly M.B., Friedrich C., Blügel S., Chulkov E.V. Quasiparticle spectrum and plasmonic excitations in the topological insulator Sb_2Te_3 . *Phys. Rev. B*, 2015, vol.91, pp.245123-8.
21. Karpinskii O. G., Shelimova L. E., Kretova M. A., Avilov E. S., Zemskov V. S. X-ray Diffraction Study of Mixed-Layer Compounds in the Pseudobinary System $\text{SnTe-Bi}_2\text{Te}_3$. *Inorganic Materials*, 2003, vol. 39, issue 3, pp. 240-246.
22. Chiu C., Hsu C., Chen S., Wu H. Phase Equilibria of the Sn-Bi-Te Ternary System. *Journal of Electronic Materials*, 2012, vol. 41, pp. 22-31.
23. Kuropatwa B. A., Kleinke H. Thermoelectric Properties of Stoichiometric Compounds in the $(\text{SnTe})_x(\text{Bi}_2\text{Te}_3)_y$ System. *Journal of Inorganic and General Chemistry*, 2012, vol. 638, issue 15, pp. 2640-2647.
24. Kuznetsova L.A., Kuznetsov V.L., & Rowe D.M. Thermoelectric Properties and Crystal Structure of Ternary Compounds in the $\text{Ge}(\text{Sn,Pb})\text{Te-Bi}_2\text{Te}_3$ Systems. *J.Phys.Chem. Solids*, 2000, vol. 61, issue 8, pp. 1269-1274.
25. Aliev Z.S., Amiraslanov I.R., Nasonova D.I., Shevelkov A.V., Abdullayev N.A., Jahangirli Z.A., Orujlu E.N., Otrokov M.M., Mamedov N.T., Babanly M.B., Chulkov E.V. Novel ternary layered manganese bismuth tellurides of the $\text{MnTe-Bi}_2\text{Te}_3$ system: Synthesis and crystal structure. *Journal of Alloys and Compounds*, 2019, vol. 789, pp. 443-450.

YENİ 9P-TİP LAYLI SnBi_4Te_4 van der VAALS BİRLƏŞMƏSİNİN SİNTEZİ VƏ KRİSTAL QURULUŞU

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Mövcud tetradimitəbənzər laylı xalkogenid fazalarının quruluş xüsusiyyətlərini araşdırmaqla SnBi_4Te_4 tərkibli qarışıq laylı birləşmənin istiqamətli sintezi həyata keçirilmiş, alınan polikristallik nümunə DTA, RFA və SEM üsulları ilə tədqiq olunmuşdur. Müəyyən edilmişdir ki, birləşmə 831 K-də peritektik reaksiya ilə parçalanmaqla əriyir. Nümunənin ovuntu rentgenoqramının Rietveld metodu ilə tədqiqi göstərir ki, birləşmə tetradimitəbənzər laylı quruluşlu olub, Bi_2 layları və SnBi_2Te_4 -laylı paketlərinin təkrarlanması ilə ibarətdir. Sn-Bi-Te üçlü sistemində belə birləşmənin olması, A^{IV} -Bi-Te ($\text{A}^{\text{IV}} = \text{Ge, Sn, Pb}$) sistemlərində də oxşar tərkibli fazaların mövcud ola biləcəyini ehtimal etməyə əsas verir.

Açar sözlər: Sn-Bi-Te üçlü sistemi, SnBi_4Te_4 , yeni laylı xalkogenid, Bi_2 layları, van der Waals birləşmələri, kristal quruluş, Rietveld metodu, topoloji izolyator.

СИНТЕЗ И КРИСТАЛЛИЧЕСКАЯ СТРУКТУРА НОВОГО СЛОИСТОГО ВАН-ДЕР-ВААЛЬСОВОГО СОЕДИНЕНИЯ SnBi_4Te_4 P-ТИПА

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Учитывая структурные особенности уже известных тетрадимитоподобных слоистых халькогенидных фаз, нами был проведен направленный синтез нового смешанно-слоистого соединения SnBi_4Te_4 . Полученный полукристаллический образец был исследован методами дифференциального термического и рентгенофазового анализом, а также сканирующей электронной микроскопии. Установлено, что полученное соединение плавится инконгруэнтно при 831 К. Из порошковой рентгенограммы методом Ритвельда определена кристаллическая структура соединения SnBi_4Te_4 выявлено, что она относится к тетрадимитному типу и характеризуется чередованием семислойных блоков SnBi_2Te_4 и бислоев висмута. Существование нового соединения SnBi_4Te_4 в тройной системе Sn–Bi–Te дает основание для поиска аналогичных фаз в других подобных системах $A^{\text{IV}}\text{–Bi–Te}$ ($A^{\text{IV}} = \text{Ge}, \text{Sn}, \text{Pb}$).

Ключевые слова: тройная система Sn–Bi–Te, SnBi_4Te_4 , новый слоистый халькогенид, Bi-бислои, ван-дер-ваальсовы соединения, кристаллическая структура, метод Ритвельда, топологический изолятор.