

UDC 546. 19.22+56.763.24

CHEMICAL INTERACTION AND GLASS FORMATION IN THE As₂S₃-CuCr₂Te₄ SYSTEM

I.I. Aliev¹, M.G. Shakhbazov², S. Sh. Ismailova¹

¹Acad. M.F. Nagiyev Institute of Catalysis and Inorganic Chemistry National Academy of Sciences of Azerbaijan e-mail: aliyevimir@rambler.ru ²Azerbaijan Pedagogical State University 34 Hajibeyov str., Baku AZ 1001

> Received 29.06.2020 Accepted 08.09.2020

The character of the chemical interaction and glass formation in the As₂S₃-CuCr₂Te₄ system was explored by means of physicochemical analysis (DTA, MSA, XRD, density and microhardness measurements) and a phase diagram was constructed. It was established that the state diagram of the system is a quasi-binary eutectic type. In the system narrow homogeneous regions of up to 1.5 mol % As 2S3 and 3.5% mol % CuCr₂Te₄ were detected. Compounds As₂S₃ and CuCr₂Te₄ with each other form a eutectic composition of 10 mol. % CuCr₂Te₄ with a melting point of 270°C. Under ordinary conditions, in the As₂S₃-CuCr₂Te₄ system based on As₂S₃, the glass-forming region reaches 15 mol % CuCr₂Te₄.

Keywords: eutectic, glass formation, density, microhardness, syngony.

DOI: 10.32737/2221-8688-2020-3-376-381

Introduction

The glassy chalcogenides As₂S₃ and alloys doped As_2Se_3 and with chalcogenides as photosensitive and magnetooptical materials have attracted the attention of researchers in recent years [1-8]. In this regard, multicomponent systems with the participation of arsenic sulfide and selenide are intensively studied [9-11]. Chromium chalcogenides and the obtained ternary compounds based on it have magnetic properties [12-14]. Therefore, the study into interaction between As₂Se₃ and CuCr₂Te₄ compounds is very important.

The aim of this work is to study chemical interactions and glass formation in the As₂S₃-CuCr₂Te₄ system and build a phase diagram.

The glassy compound As₂S₃ melts with open maximum at 310°C and crystallizes in monoclinic syngony with lattice parameters as follows: a = 11.49; b = 9.59; c = 4.25 Å, $\beta =$ $90^{\circ}27'$, sp.gr. P 21/n [15]. The density and microhardness of the glassy As₂S₃ compound $\rho = 3.187 \text{ g/cm}^3 \text{ and } H\mu =$ are respectively 128-145 MPa.

The initial compound CuCr₂Te₄ melts congruently at 1155°C [14] and crystallizes in a cubic syngony with lattice parameters a =11.134 Å, sp.gr. Fd-3m, density $\rho = 6.51 \text{ g/cm}^3$ [16].

Experimental part

The alloys were annealed at 270°C for 350 hours. Alloys of the system were obtained from ligatures As2S3 and CuCr2Te4, previously synthesized from elements in quartz ampoules pumped out to 0.133 Pa in the range of temperature 600-1200°C depending on the alloy composition. The alloys were annealed at 270°C for 350 hours.

The alloys of the As₂S₃-CuCr₂Te₄ system were studied by differential thermal (DTA), X-ray diffraction (XRD), microstructural (MSA) analysis, as well as microhardness measurements and determination.

Thermal analysis of the alloys was carried out on a TERMOSKAN-2 instrument with accuracy of 3-5°C, whiht chromel-alumel thermocouple. The heating rate is 5 deg/min. X-ray phase analysis was performed on an X-ray device of D2 PHASER model using CuKα radiation and a Ni filter. Microstructural analysis of alloys of the As2S3-CuCr2Te4

system was carried out by means of a MIM-8 microscope. The microhardness of the phases was measured on a PMT-3 instrument with accuracy of 5% while the density of the samples was determined by the pycnometric method.

Results and its discussion

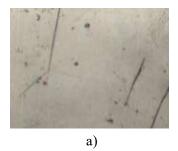
Alloys of the As_2S_3 -CuCr $_2Te_4$ system with high As_2S_3 content are bright red, brittle layered, and with the CuCr $_2Te_4$ increased content; the samples acquire a dark brown hue. Alloys rich in As_2S_3 are obtained by a glassy appearance. They are resistant to air and water. Concentrated mineral acids (HNO $_3$, H $_2SO_4$) and alkalis (NaOH, KOH) decompose them. In

order to crystallize glassy alloys, 0-25 mol % of As₂S₃-based CuCr₂Te₄ was annealed at 170° C for 600 hours while the remaining alloys were annealed at 300°C for 240 hours. A physicochemical study was performed before and after annealing. Some physicochemical properties of alloys from the glass region are given in Table 1.

Table 1. Some physicochemical properties of glasses of the As ₂ S ₃ -CuCr ₂ Te ₄ system						
mposition,	Thermal effects, °C	Microhardness				

Composition, mol %		Thermal effects, °C			Microhardness, MPa	Density q/cm ³	Results MSA	
As_2S_3	CuCr ₂ Te ₄	Tg	Ткр.	Тпл.	. Wira			
100	0	170	230	310	1350	3,20	Glass,	
97	3	170	230	210	1380	3,25	_	
95	5	175	240	315	1380	3,32	_	
93	7	180	245	320	1380	3,39	_	
90	10	190	250	275	1390	3,47	_	
85	15	195	255	320	1400	3,58		
80	20	200	260	330	1420	3,70	_	
75	25	210	270	340	1430	3,83	Glass, crystal.	
70	30	210	275	350	1450	3,96	Glass, crystal.	
60	40	220	280	350	1450	4,25	Glass, crystal.	

Thermal analysis of alloys of the As_2S_3 -Cu Cr_2Te_4 system showed that thermograms of alloys are observed two endothermic effects related to solidus and three endothermic effects related to liquidus. DTA of the system alloys before annealing shows that on thermograms of the alloys in the concentration range of 0–25 mol. % Cu Cr_2Te_4 , a softening temperature is observed at Tg of 170°C. After prolonged heat treatment at 170 ° C for 600 h, the alloys crystallize and a series of effects is observed on the thermograms.



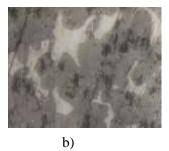




Fig. 1. Microstructure of alloys of the As₂S₃-CuCr₂Te₄ system. a) - 1.5; b) -90.; c) - 96.5 mol % CuCr₂Te₄.

Microstructural analysis of the alloys of the As_2S_3 -CuCr $_2Te_4$ system shows that, in addition to the initial components, the remaining alloys are biphasic. The microstructure of alloys from the region of 0-15 mol % CuCr $_2Te_4$ represents one phase of dark color, and in the concentration range from 15 to 25 mol. % CuCr $_2Te_4$ crystalline inclusions appears. A study of the microstructure of the annealed samples showed that 1.5 mol % CuCr $_2Te_4$ and 3.5 mol. % As_2S_3 single-phase alloys (Fig. 1a and c). As the content of CuCr $_2Te_4$ rises to 10 mol. % As_2S_3 solid solutions disintegrate and the alloys become biphasic (Fig. 1 b)

RAF of unannealed alloys of the As2S3-CuCr2Te4 system shows that only on thermograms of cast samples with a content of more than 25 mol. % CuCr2Te4, intense diffraction lines are observed. After annealing at 170°C for 600 hours on diffraction patterns in

the concentration range of 0-25 mol intense diffraction maxima CuCr₂Te₄ obtained. X-ray analysis showed that the diffraction patterns of alloys 70 and 100 mol % CuCr₂Te₄ consist of mixed diffraction lines of the starting components. These data confirm that the alloys of the system are two-phased. Fig. 2 shows diffraction patterns of alloys in the system of compositions 10, 15, 25, 70, and 100 mol % CuCr₂Te₄. As seen in Fig. 1, alloys 10, 15 mol % CuCr2Te4 belongs to the field of glasses, 25 mol % CuCr2Te4 in the glasscrystalline region, 70 and 100 mol % CuCr₂Te₄ crystalline region.

DTA, XRD and MSA showed that in the system with slow cooling of the samples, the region of glass formation reaches 15 mol % CuCr₂Te₄, as well as in the regime of quenching in ice water of about 20 mol % CuCr₂Te₄.

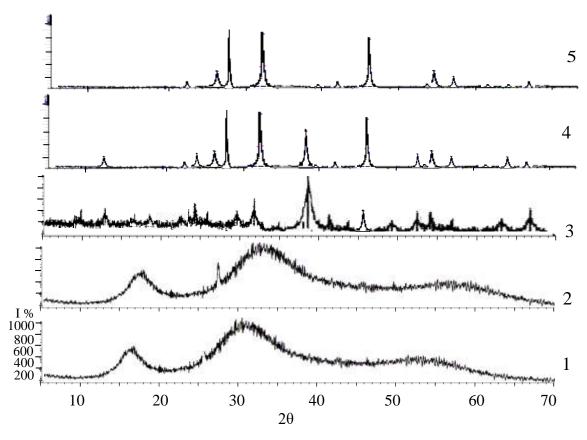


Fig. 2. Diffraction patterns of alloys of the As₂S₃-CuCr₂Te₄ system. 1-10, 2-15, 3-25, 4-70, 5-100 mol % CuCr₂Te₄.

The T-x phase diagram of the As_2S_3 -CuCr₂Te₄ system built in line with the physicochemical

analysis is shown in Fig. 3. The state diagram of the As_2S_3 -CuCr₂Te₄ system is a quasi-binary eutectic type (Fig. 3).

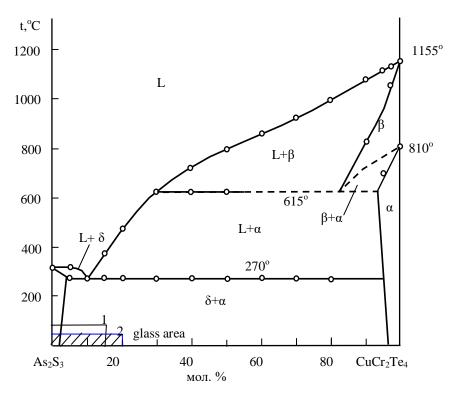


Fig. 3. T-x phase diagram of the As₂S₃-CuCr₂Te₄ system. Area of glass formation: 1-slow cooling, 2-quenching in ice water.

The liquidus of the system consists of monovariant curves of δ , α and β -phase. The phases δ and α form an eutectic with composition of 10 mol % CuCr₂Te₄ and melt at 270 ° C. At the eutectic point, a three-phase reaction occurs: $\mathfrak{K} \leftrightarrow \delta + \alpha$. In the range of 1.5-

96 mol % CuCr₂Te₄ concentrations below the solidus line, two-phase alloys crystallize $(\delta + \alpha)$. In the range of 96.5-100 mol % CuCr₂Te₄ crystallizes α phase. The microhardness of As₂S₃-CuCr₂Te₄ alloys was measured at each phase before and after annealing.

Table 2. Results of DTA measurements of microhardness and density of system alloys As₂S₃-CuCr₂Te₄ after annealing

Composition, mol %			Density, g/cm ³	Microhardness, MPa	
		Thermal effects, °C		α	В
As_2S_3	CuCr ₂ Te ₄	Thermal effects, C			
2-22-3	0 0 0 2 2 0 0 4				
				P=0,10 H	P=0,15 H
100	0.0	310	3,46	700	-
95	5,0	270,310	3,60	750	-
90	10	270	3,76	Eutec.	Eutec.
85	15	270,375	3,92	-	-
80	20	270,480	4,07	800	-
70	30	270,615	4,37	800	-
60	40	270,615,720	4,68	-	-
50	50	270, 615,800	4,98	-	-
40	60	270,860	5,28	-	1870
30	70	270,925	5,60	-	1870
20	80	270,1000	5,90	-	1880
10	90	830,1075	6,21	-	1880

ш						
	5,0	95	700,1120	6,57	-	1880
	3,0	97	1130	6,54	-	1860
	0,0	100	810, 1155	6,51	-	1850

When measuring the microhardness of the system alloys, three series of values were obtained (Table 1). Before annealing, the microhardness of glassy alloys 0-10 mol % $CuCr_2Te_4$ in the range of microhardness varies in the range (1350-1400) MPa. After annealing the same alloys, the microhardness decreases (700-800) MPa. The microhardness of α phase (CuCr₂Te₄-based solid solutions) varies in the range (1850-1880) MPa. Prior to annealing, the density values of glassy alloys in the

concentration range of 0–20 mol % CuCr₂Te₄ vary in the range of 3.20–3.70 g/cm³ (Table 2). After annealing, the density values of the alloys in the same region vary within 3.46-3.93 g/cm³ (Table 2). The data obtained show that the microhardness of glasses is higher than the corresponding crystals. On the contrary, the density of glasses is lower than that of the corresponding crystals, which is in good agreement with the literature.

References

- 1. Babayev A.A., Muradov R., Sultanov S.B., Askhabov A.M. Influence of the preparation conditions on the optical and photoluminescent properties of glassy As₂S₃. *Inorganic Materials*. 2008, vol. 44, no. 11, pp. 1187-1201.
- 2. Verlan V.I. Native cenyers of elektron and hole traps in thin amorphous films As₂S₃ AND As₂Se₃. *Journal of Optoelectronics and Advanced Materials*. 2003, vol. 5, no. 5, pp. 1121-1134.
- 3. Seeva Kandpal, Kushwaha R. P. S. Photoacoustic spectroscopy of thin films of As₂S₃, As₂Se₃ and GeSe₂. *Indian Academy of Sciences. PRAM ANA journal of physics*. 2007, vol. 69, no. 3, pp. 481-484.
- 4. Littler I.C.M., Fu L.B., Mägi E.C., Pudo D., Eggleton B.J. Widely tunable, acoustooptic resonances in Chalcogenide As₂Se₃ fiber. *Optics Express.* 2006, vol.14, issue 18, pp. 8088-8095.
- 5. Cao H., Xiao Y., Lu Y. et al. Ag₂Se complex nanostructures with photocatalytic activity and super hydrophobicity. *Nano Res.*, 2010, vol. 3, no. 12, pp. 863-873. DOI: 10.1007/s 12274-010-0057.
- 6. Ashok U., Mitkari V., Choudhari D. Et al. Synthesis of nanostructured Cu: As₂S₃ thin films by chemical bath deposition method and their physical properties. *Int. Journal of Materials and Chemistry*. 2013, vol. 2, no. 3, pp. 33-38.
 - DOI: 10.5923/j.ijmc.20130302.03.
- 7. Andriesh A.M., Verlan V.I. Donor- and

- acceptor-like center revealing by Photoconduktivity of amorphous thin As_2Se_3 films. *Journal of Optoelectronic and Advanced Materials*. 2001, vol. 3, no. 2, pp. 455 458.
- 8. Lovu M., Shutov S., Rebeja S., Colomeco E., Popescu M. Effect of material additives on photodarkening in amorphous As₂Se₃ films. *Journal of Optoelec. and Advanced Materials.* 2000, vol. 2, no. 1, pp. 53-58.
- 9. Aliyev I.I., Hasanguliyeva Sh.A., Ilyasli T.M. Phase equilibrium and glass formation in the AsSe-MnAs₂Se₄ system. *Chemical problems*. 2014, no.2, pp. 215-218.
- 10. Aliyev I.I., Ilyasly T.M., Hasangulieva Sh.A., Veliyev J.A. Phase equilibria and glass formation in the AsSe-MnSe system. *Inorganic Materials*. 2011, vol. 47, no. 7, pp. 784-787.
- 11. Busheva E.V., Shabunina G., Aminov T.G. A Study of Interaction in the As₂Se₃–Cr₂Se₃ System. *Russ. J. Inorg. Chem.* 1999, vol. 44, no. 6, pp. 922-925.
- 12. Бабицына А.А., Конешова Т.И., Калинников В.Т. Исследование возможности образования твердых растворов в системах CuCr₂Se₄-InSe; CuCr₂Se₄-In₂Se₃; CuCr₂Se₄-CuInSe₂. *Inorganic Materials*. 1981, vol. 17, no. 9, p.1716.
- 13. Бабицына А.А., Емельянова Т.А., Конешова Т.И. Взаимодействие в системе Си-Сг- Те. *Russ. J. Inorg. Chem.*, 2000, vol. 45, no. 8, pp.1397-1400.

- 14. Yamashita O., Yamauchi H., Yamaguchi Y . et al. Magnetic Properties of the System CuCr₂Se_{4-x}Y_x (Y=C1, Br). *J. Phys. Soc. Jap.* 1979, vol. 47, no. 2, pp. 450-454.
- 15. Khvorostenko A.S. Arsenic chalcogenides. Review from the series Physical and Chemical Properties of Solids. 1971, 93 p.
- (In Russian).
- 16. Riedel E., Horvath E.Z. Roentgeno graphische Untersuchund der systeme CuCr₂ (S_{1-x} Se_x)₄ und CuCr₂ (Se_{1-x}Te_x)₄. *Anorg. Allg. Chem.* 1973, vol. 399, pp. 219-223.

As₂S₃-CuCr₂Te₄ SİSTEMİNDƏ KİMYƏVİ QARŞILIQLI TƏSİR VƏ ŞÜŞƏƏMƏLƏGƏLMƏ

İ.İ. Əliyev, M.G. Şahbazov, S. Ş. İsmayılova

AMEA-nın akad. M.Nağıyev adına Kataliz və Qeyri-üzvi Kimya İnstitutu AZ 1143, Bakı, H.Cavid pr., 113; e-mail: <u>aliyevimir@rambler.ru</u>
Azərbaycan Dövlət Pedaqoji universiteti
AZ 1001 Bakı, Ü.Hacıbəyov küç.34

 As_2S_3 -Cu Cr_2Te_4 sistemində kimyəvi qarşılıqlı təsir və şüşəəmləgəlmənin xüsusiyyətləri fiziki-kimyəvi analiz metodları (DTA, MSA, XRD, həmçinin sıxlıq və mikrobərkliyin ölçmələri) ilə tədqiq edilmiş və faza diaqramı qurulmuşdur. Nəticədə müəyyən edilmişdir ki, sistemin hal diaqramı kvazibinar olub, evtektik tiplidir. Sistemdə As_2S_3 yaxınlığındakı, 1.5 mol % -ə qədər məhdud sahədə bərk məhlul sahəsi mövcuddur, Cu Cr_2Te_4 birləşməsi əsasında isə 3.5 mol % bərk məhlul aşkar edilmişdir. As_2S_3 və $CuCr_2Te_4$ birləşmələri arasında əmələ gələn evtektikanın tərkibi 10 mol % $CuCr_2Te_4$ və əriməsi isə 270°C-dir. As_2S_3 -Cu Cr_2Te_4 sistemində adi soyudulma şəraitində, As_2S_3 əsasında şüşə sahəsi 15 mol % $CuCr_2Te_4$ təşkil edir.

Açar sözlər: evtektik, şüşəəmələgəlmə, sıxlıq, mikrobərklik, sinqoniya.

ХИМИЧЕСКОЕ ВЗАИМОДЕЙСТВИЕ И СТЕКЛООБРАЗОВАНИЕ В СИСТЕМЕ As₂S₃-CuCr₂Te₄

И.И. Алиев, М.Г. Шахбазов, С.Ш. Исмаилова

Институт катализа и неорганической химии им. акад. М.Нагиева Национальной АН Азербайджана AZ 1143 Баку, пр. Г.Джавида, 113; e-mail:aliyevimir@rambler.ru Азербайджанский Государственный Педагогический университет AZ 1001 Баку, ул.У. Гаджибекова, 34

Характер химического взаимодействия и стеклообразования в системе As_2S_3 - $CuCr_2Te_4$ исследован методами физико-химического анализа (ДТА, МСА, РФА а также измерением плотности и микротвердости) и построена фазовая диаграмма. Установлено, что диаграмма состояния системы является квазибинарной эвтектического типа. В системе выявлены узкие гомогенные области до 1.5 мол. % As_2S_3 и 3.5 мол. %. $CuCr_2Te_4$. Соединения As_2S_3 и $CuCr_2Te_4$ между собой образуют эвтектику состава 10 мол. % $CuCr_2Te_4$ с температурой плавления 270° С. При обычных условиях охлаждения в системе As_2S_3 - $CuCr_2Te_4$ на основе As_2S_3 область стеклообразования доходит до 15 мол. % $CuCr_2Te_4$.

Ключевые слова: эвтектика, стеклообразование, плотность, микротвердость, сингония.